Amphidinolide W, a New 12-Membered Macrolide from Dinoflagellate *Amphidinium* sp.

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Abstract: A new cytotoxic 12-membered macrolide, amphidinolide W (1), has been isolated from a marine dinoflagellate *Amphidinium* sp., and the structure was elucidated by spectroscopic data including $^{13}C^{-13}C$ INADEQUATE correlations for its ^{13}C -enriched sample. The absolute stereochemistry of 1 was assigned by combination of *J*-based configuration analysis and modified Mosher method. Amphidinolide W (1) is the first macrolide without an exomethylene unit among all amphidinolides obtained so far.

Amphidinolides are a series of unique macrolides obtained from marine dinoflagellates of the genus Amphidinium, which are symbionts of Okinawan marine acoel flatworms Amphiscolops spp.1 Our continuing search for bioactive secondary metabolites from laboratory-cultured marine dinoflagellates² resulted in the isolation of a new cytotoxic 12-membered macrolide, amphidinolide W (1), from extracts of the strain (Y-42) of the dinoflagellate Amphidinium sp. The gross structure of **1** was elucidated on the basis of the spectroscopic data including ¹³C⁻¹³C correlations obtained from the INADEQUATE spectrum. The absolute configurations at C-11, C-12, and C-14 of 1 were assigned by J-based configuration analysis and a modified Mosher method, while those at C-2 and C-6 were elucidated from NMR data of MTPA esters of a reductive product of 1. Here, we describe the isolation and structure elucidation of 1.



The dinoflagellate *Amphidinium* sp. (strain Y-42) was obtained from the inside cells of a marine acoel flatworm *Amphiscolops* sp. collected off Sunabe, Okinawa. The dinoflagellate was mass cultured unialgally at 25 °C for

Table 1.¹H and ¹³C NMR Data of Amphidinolide W (1)in CDCl3

		III CDCI3	
position	$\delta_{\rm C}$	$\delta_{ m H}$ (m, Hz)	HMBC (H)
1	175.3 s		H-3b ^c , H-11, H ₃ -21
2	39.4 d	2.64 m	H-3b ^c , H ₂ -4, H ₃ -21
3	25.8 t	2.17 m	H ₂ -4, H ₃ -21
		1.65 m	
4	35.9 t	2.49 dt, 6.3, 18.3	H-2, H-3b ^c
		2.34 m	
5	212.8 s		H-3b ^c , H ₂ -4,
			H-6, H ₃ -22
6	45.8 d	2.38 m	H-8a ^c , H ₃ -22
7	32.3 t	1.89 m	H-6, H-8b ^c , H ₃ -22
		1.50 m	
8	32.3 t	2.31	H-7a ^c , H-9, H-10
		1.89 m	
9	138.2 d	5.64 ddd, 5.5, 9.7, 15.5	H-7b ^c , H-8b ^c ,
			H-10, H-11
10	127.0 d	5.52 dd, 8.3, 15.5	H-8b ^c , H-9, H-11
11	79.0 d	4.96 dd, 6.5, 8.3	H-9, H-10
12	70.6 d	3.56 ddd, 2.2, 6.5, 10.0	H-11, H-13a ^c
13	41.0 t	1.41 ddd, 3.3, 10.0, 13.9	H ₃ -23
		1.26 ddd, 2.2, 11.5, 13.9	
14	28.8 d	2.83 m	H-12, H ₂ -13, H ₃ -23
15	135.6 d	5.04 d, 10.0	H ₂ -13, H ₃ -23, H ₃ -24
16	133.5 s		H-15, H ₃ -24
17	133.6 d	6.03 d, 15.5	H-15, H-18,
			H ₂ -19, H ₃ -24
18	129.8 d	5.62 ddd, 6.8, 8.6, 15.5	H ₂ -19, H ₃ -20
19	25.6 t	2.11 ^a m	H ₃ -20
20	13.8 q	1.02^{b} t, 7.2	H-18, H ₂ -19
21	16.4 q	1.15 ^{<i>b</i>} d, 6.7	
22	18.6 q	1.03^{b} d, 7.2	
23	21.7 q	0.96 ^b d, 6.7	H-13b, H-15
24	12.7 q	1.77^{b} s	

^{*a*} 2H. ^{*b*} 3H. ^{*c*} a and b denote low-field and high-field resonances, respectively, of a geminal pair for C-3, C-7, C-8, and C-13.

14 days in a seawater medium enriched with 1.2 mM of 13 C-labeled sodium bicarbonate and 1% ES supplement. The harvested algal cells (35.9 g, wet weight, from 40 L of culture) were extracted with MeOH/toluene (3:1), and the extracts were partitioned between toluene and water. The toluene-soluble materials were subjected to silica gel column chromatography (CHCl₃/MeOH) followed by C₁₈ HPLC (CH₃CN/H₂O) to afford amphidinolide W (**1**, 0.009%, wet weight) together with two known macrolides, amphidinolides G³ (0.0008%) and H³ (0.0007%).

FABMS of amphidinolide W (1) showed a pseudomolecular ion peak at m/z 413 (M + Na)⁺, and 1 had a molecular formula of C₂₄H₃₈O₄ as revealed by HRFABMS [m/z 413.2686 (M + Na)⁺, +1.8 mmu]. IR absorptions at 3442 and 1721 cm⁻¹ indicated the presence of hydroxyl-(s) and carbonyl group(s), respectively. The UV spectrum showed the absorption at 235 nm (ϵ 14 000), implying the presence of a conjugated diene chromophore. In the ¹H NMR spectrum of 1, intense satellite signals due to ¹³Cenrichments were observed for all proton signals. The average incorporation ratio of carbon-13 in each carbon atom was estimated to be ca. 30% on the basis of intensity of proton satellite signals. ¹H and ¹³C NMR data (Table 1) disclosed the existence of a ketone, an ester carbonyl, an sp² quaternary carbon, five sp² methines, five sp³

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Figure 1. Selected 2D NMR correlations for amphidinolide W (1).



Figure 2. Rotation model for C-11-C-12 bond of amphidinolide W (1). NOESY correlations are illustrated by solid arrows.

methines (two of them bearing an oxygen atom), six sp³ methylenes, and five methyls (one of them attached to an olefin). Since five out of six unsaturations were accounted for, amphidinolide W (1) was inferred to contain one ring. Detailed analyses of the ${}^{1}H-{}^{1}H COSY$, TOCSY, and HMQC spectra disclosed three proton networks from H₃-21 to H₂-4, from H₃-22 to H₃-24, and from H-17 to H₃-20 (Figure 1). HMBC correlations of H₂-4 and H₃-22 to a ketone carbonyl carbon ($\delta_{\rm C}$ 212.8, C-5) suggested that C-4 and C-6 were linked through the ketone carbonyl. The presence of a conjugated diene moiety at C-15-C-18 was deduced from HMBC correlations of H-17 to C-15 and H₃-24 to C-15 and C-16. The carbon chain from C-1 to C-20 containing four branched methyl groups was also supported by one-bond ¹³C-¹³C correlations observed in the INADEQUATE spectrum. The existence of an ester linkage between C-1 and C-11 was implied by HMBC correlations of H₃-21 and H-11 to C-1 ($\delta_{\rm C}$ 175.3). Geometries of two disubstituted olefins at C-9–C-10 and C-17–C-18 were assigned as both *E* by $^{1}\text{H}^{-1}\text{H}$ coupling constants [J(H-9/H-10); 15.5 Hz, J(H-17/H-18); 15.5 Hz], while *E*-geometry of a trisubstituted olefin at C-15-C-16 was revealed by NOESY cross-peaks for H-14/H₃-24 and H-15/H-17. Thus, the gross structure of amphidinolide W was elucidated to be 1.

The relative configurations at C-11, C-12, and C-14 were elucidated on the basis of ¹H-¹H and ¹³C-¹H coupling constants⁴ in addition to NOESY correlations. For the C-11–C-12 bond (Figure 2), the ³*J*(H-11, H-12) (6.5 Hz) suggested that this bond underwent a conformational change between anti and gauche. Furthermore, the ${}^{2}J(C-12, H-11)$ showed a medium value (-2.0 Hz), also implying that H-11 had both anti and gauche relations to OH-12 due to rotation of the bond. The values for ${}^{3}J(C-13, H-11)$ (~0 Hz) and ${}^{3}J(C-10, H-12)$ (2.3 Hz) obtained from the hetero half-filtered TOCSY (HETLOC)⁵ spectrum indicated that H-11 and H-12 were gauche to C-13 and C-10, respectively. Of the six possible rotamers



Figure 3. $\Delta \delta$ values $[\Delta \delta$ (in ppm) = $\delta_S - \delta_R$] obtained for (S)- and (R)-MTPA esters (2a and 2b, respectively) of amphidinolide W (1).

arising from erythro and threo, only one pair of threo relationship shown in Figure 2 satisfied all of these coupling data. NOESY correlations also supported the threo relation for the C-11-C-12 bond. For the C-12-C-13 bond, ³J(H-12, H-13b) was a typical value (2.2 Hz) for gauche relationships, while ³J(H-12, H-13a) (10.0 Hz) was indicative of an anti relation for H-12-H-13a. Combination of two-bond ¹³C-¹H coupling constants of C-12/H-13a (-5.0 Hz) and C-12/H-13b (~0 Hz) with NOESY correlations for H-11/H-13a and H-11/H-13b suggested that H-13a was gauche to C-11 and 12-OH and that H-13b was gauche to C-11 and anti to 12-OH. On the other hand, a rotation model for the C-13-C-14 bond was deduced from the following coupling constants: ³*J*(H-13a, H-14) 3.3 Hz, ³J(H-13b, H-14) 11.5 Hz, ³J(C-15, H-13a) 7.6 Hz, ³J(C-15, H-13b) 2.2 Hz, ³J(C-23, H-13a) ~0 Hz, and ³J(C-23, H-13b) 0.9 Hz. This was also supported by NOESY correlations for H-12/H-15, H₂-13/H₃-23, and H-13a/H-14. Therefore, the 1,3-chiral center of C-12-C-14 was elucidated to have a syn relation.

The absolute configuration at C-12 of amphidinolide W (1) was determined by a modified Mosher method.⁶ Treatment of 1 with (R)-(-)- and (S)-(+)-2-methoxy-2trifluoromethyl-2-phenylacetyl chloride (MTPACl) afforded the (S)- and (R)-MTPA esters (2a and 2b, respectively). $\Delta\delta$ values ($\delta_S - \delta_R$) of H-9, H-10, and H-11 showed positve values, while those of H₂-13, H-14, H-15, and H₃-23 were negative (Figure 3), suggesting that C-12 possessed S-configuration. Thus, the absolute configurations at C-11 and C-14 were elucidated to be S and R, respectively. The absolute configuration at C-2 was investigated by NMR data of MTPA esters of reduction products of 1 as follows. Treatment of 1 with LiAlH₄ and then (R)- or (S)-MTPACl afforded tetrakis-(S)- and (R)-MTPA esters (3a and 3b, respectively) of the linear alcohol of 1 (Scheme 1). The methylene protons of C-1 for 3a appeared as separated double doublet signals at $\delta_{\rm H}$ 3.94 and 4.12, while those for **3b** were observed as overlapped 2H signals at $\delta_{\rm H}$ 4.09, suggesting *S*-configuration at C-2.⁷ To determine the absolute configuration at C-6 of 1, Baeyer-Villiger reaction was applied to obtain the segment including the methine carbon at C-6. Amphidinolide W (1) was reduced with hydrogen gas and

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⁽⁷⁾ Mosher's method has been applied for determination of absolute stereochemistry of a methyl group at C-25 of steroids with a primary hydroxy group at C-26. In the ¹H NMR spectra of the (+)-(\vec{R})-MTPÅ esters, two 26-methylene protons of the 25S isomer are much closer $(\Delta \delta \text{ ca. } 0.04)$ to each other than those $(\Delta \delta \text{ ca. } 0.14)$ of the 25*R* isomer, whereas in the (-)-(S)-MTPA esters, the mutual relation is reverse. (a) De Riccardis, F.; Minale, L.; Riccio, R.; Giovannitti, B.; Iorizzi, M.; Debitus, C. *Gazz. Chim. Ital.* **1993**, *123*, 79–86. (b) Finamore, E.; Minale, L.; Riccio, R.; Rinaldo, G.; Zollo, F. J. Org. Chem. 1991, 56, 1146-1153.



Figure 4. $\Delta \delta$ values [$\Delta \delta$ (in ppm) = $\delta_S - \delta_R$] obtained for tris-(*S*)- and tris-(*R*)-MTPA esters (**4a** and **4b**, respectively) of the C-6–C-20 segment of amphidinolide W (**1**).



rhodium—aluminum to avoid epoxidation of three olfins. Baeyer—Villiger reaction was performed by treatment of the crude hexahydro form of **1** with TFPA (trifluoroperoxyacetic acid). After reduction with LiAlH₄, esterification with (R)—(–)- and (S)—(+)-MTPACl and HPLC separation afforded the tris-(*S*)- and tris-(*R*)-MTPA esters (**4a** and **4b**, respectively) of the C-6—C-20 segment of **1**. $\Delta\delta$ values obtained from the ¹H chemical shifts of **4a** and **4b** were shown in Figure 4. The $\Delta\delta$ value of H₃-22 (–0.08) showed a negative sign, while those of H₂-7 (+0.04 and +0.03) and H₂-8 (+0.01) were positive, suggesting 6*S*configuration. Therefore, the absolute configurations of five chiral centers in amphidinolide W (**1**) were elucidated to be 2*S*, 6*S*, 11*S*, 12*S*, and 14*R*.

Amphidinolide W (1) is the first macrolide without an exomethylene unit among all of the amphidinolides isolated so far. The gross structure of C-9–C-16 moiety of amphidinolide W (1) corresponds to that of C-6–C-15 of amphidinolide H, which was contained in this strain Y-42, suggesting that amphidinolide W (1) may be biogenetically related to amphidinolide H. Amphidinolide W (1) exhibited cytotoxicity against murine lymphoma L1210 cells in vitro with an IC₅₀ value of 3.9 μ g/mL.

Experimental Section

General Methods. ¹H and ¹³C NMR spectra were recorded on a 600 MHz spectrometer using 2.5 mm micro cells for CDCl₃ (Shigemi Co., Ltd.). 2D NMR spectra were measured on a 500 MHz spectrometer using 5 mm micro cells for CDCl₃ (Shigemi Co., Ltd.). HETLOC experiments were performed using the pulse sequence proposed by Woolborn and Leibfritz with composite pulses for broadband constant rotations (bandwidth ±0.60).⁸ The duration of the trim pulse, the delay in the BIRD pulse, and the constant time for *J*_{CH} evaluation were 2.5, 300, and 3.57 ms, respectively. The MLEV17 spin-lock period was set to 30 ms for ^{2.3}*J*_{C,H}. For 256 *t*₁ increments, 256 transients with 16 dummy scans were accumulated in 1 K data points. Zero-filling to 1 K for *F*₁ and multiplication with squared cosine-bell windows shifted in both dimensions were performed prior to 2D Fourier transformation. The measuring time was ca. 48 h. FABMS spectra were recorded using p-nitrobenzyl alcohol containing sodium iodide as a matrix in positive mode.

Cultivation and Isolation. The dinoflagellate Amphidinium sp. (strain number Y-42) was separated from the inside cells of a marine acoel flatworm Amphiscolops sp., which was collected off Sunabe, Okinawa. The dinoflagellate was unialgally cultured at 25 °C for 2 weeks in seawater medium enriched with NaH13-CO₃ (1.2 mM) and 1% ES supplement. The harvested cells (36 g, wet weight, from 40 L of culture) were extracted with MeOH/ toluene (3:1, 150 mL \times 3). After addition of 1 M NaCl (aq) (150 mL), the mixture was extracted with toluene (150 mL \times 3). The toluene-soluble fractions were evaporated under reduced pressure to give a residue (778 mg), which was subjected to a silica gel column (CHCl₃/MeOH, 98:2) and then a Sep-Pak cartridge C₁₈ (MeOH/H₂O, 8:2). The fraction containing macrolides was purified by C₁₈ HPLC [Mightysil RP-18, 5 μ m, Kanto Chemical Co., Inc., 10×250 mm; eluent, CH₃CN/H₂O (85:15); flow rate, 3 mL/min; UV detection at 220 nm] to afford amphidinolide W (1, 3.3 mg, 0.009%, wet weight, $t_{\rm R}$ 15.4 min) together with amphidinolides G (0.0008%) and H (0.0007%).

Amphidinolide W (1): colorless oil; UV (EtOH) λ_{max} 235 nm (ϵ 14 000); IR (KBr) ν_{max} 3442, 2962, 2927, 1721 cm⁻¹; ¹H and ¹³C NMR (Table S1, Supporting Information); FABMS *m*/*z* 413 (M + Na)⁺; HRFABMS *m*/*z* 413.2686 [calcd for C₂₄H₃₈O₄Na (M + Na)⁺, 413.2668].

(S)-MTPA Ester (2a) of Amphidinolide W (1). To a CH₂- Cl_2 solution (50 μ L) of amphidinolide W (1, 0.5 mg) was added 4-(dimethylamino)pyridine (50 μ g), triethylamine (5 μ L), and (R)-(–)-MTPACl (1 μ L) at 4 °C, and stirring was continued for 3 h. After addition of N, N-dimethyl-1,3-propanediamine (5 μ L) and evaporation of solvent, the residue was passed through a silica gel column (hexane/EtOAc, 5:1) to afford the (S)-MTPA ester (2a, 0.5 mg) of 1. 2a: colorless oil; ¹H NMR (CDCl₃) δ 0.89 (6H, m), 1.02 (3H, d, J = 6.7 Hz), 1.03 (3H, t, J = 7.0 Hz), 1.42–1.50 (2H, m), 1.55-1.65 (2H, m), 1.67 (3H, s), 1.75-1.90 (2H, m), 2.11 (2H, m), 2.22 (1H, m), 2.29-2.36 (3H, m), 2.45-2.50 (3H, m), 3.54 (3H, s), 5.00 (1H, d, J = 10.0 Hz), 5.21 (1H, m), 5.33 (1H, m), 5.34 (1H, m), 5.63 (1H, ddd, J = 6.5, 8,7, 15.5 Hz), 5.69 (1H, m), 6.00 (1H, d, J = 15.5 Hz), 7.35-7.43 (3H, m), and 7.59 (2H, m): ESIMS m/z 629 (M + Na)⁺: HRESIMS m/z 629.3048 [calcd for $C_{34}H_{45}O_6F_3Na (M + Na)^+$, 629.3066].

(*R*)-MTPA Ester (2b) of Amphidinolide W (1). Amphidinolide W (1, 0.5 mg) was treated with (*S*)-(+)-MTPACl (1 μ L) by the same procedure as described above to afford the (*R*)-MTPA ester (2b, 0.3 mg) of 1. 2b: colorless oil;¹H NMR (CDCl₃) δ 0.93 (3H, d, J = 6.7 Hz), 0.99–1.04 (9H, m), 1.40–1.58 (4H, m), 1.63 (3H, s), 1.72 (1H, m), 1.86 (1H, m), 2.12 (2H, m), 2.17–2.35 (4H, m), 2.44 (1H, m), 2.46–2.55 (2H, m), 3.52 (3H, s), 5.03 (1H, d, J = 10.2 Hz), 5.18 (1H, m), 5.27 (1H, m), 5.33 (1H, m), 5.59–5.67 (2H, m), 6.01 (1H, d, J = 15.4 Hz), 7.37–7.43 (3H, m), and 7.60 (2H, m); ESIMS m/z 629 (M + Na)⁺; HRESIMS m/z 629.3053 [calcd for C₃₄H₄₅O₆F₃Na (M + Na)⁺, 629.3066].

Tetrakis-(S)-MTPA Ester (3a) of Reduction Product of Amphidinolide W (1). Amphidinolide W (1, 0.5 mg) in THF solution (50 μ L) was treated with LiAlH₄ (1.5 mg) at 4 °C for 100 min. After addition of 1 M phosphate buffer (pH 6.85) (100 μ L), the mixture was extracted with EtOAc (200 μ L imes 3). The mixture, after evaporation, was dissolved in CH_2Cl_2 (50 μ L) and treated with 4-(dimethylamino)pyridine (50 μ g), triethylamine (5 μ L), and (*R*)-(–)-MTPACl (3 μ L) at 4 °C for 14 h. After addition of N,N-dimethyl-1,3-propanediamine (5 µL) and then evaporation of the solvent, the residue was subjected to SiO₂ column chromatography (hexane/EtOAc) followed by C18 HPLC (TSKgel ODS-100S, TOSOH Co., Inc., 4.6 \times 250 mm; eluent, CH₃-CN/H₂O (95:5); flow rate, 1 mL/min; UV detection at 230 nm) to afford tetrakis-(S)-MTPA ester (3a, 0.4 mg, $t_{\rm R}$ 28.5 min) of reduction product of amphidinolide W (1). 3a: colorless oil;1H NMR (CDCl₃) δ 0.72 (1.5H, d, J = 6.6 Hz), 0.79 (1.5H, d, J =6.6 Hz), 0.85 (1.5H, d, J = 6.7 Hz), 0.86 (1.5H, d, J = 6.7 Hz), 0.89 (3H, d, J = 7.2 Hz), 1.02 (3H, t, J = 7.0 Hz), 1.40-1.65 (4H, m), 1.62 (3H, s), 1.63 (1H, m), 1.76 (1H, m), 1.90-2.03 (2H, m), 2.10 (2H, m), 2.27-2.36 (2H, m), 2.45 (1H, m), 3.46 (1.5H, s), 3.47 (3H, s), 3.48 (3H, s), 3.49 (1.5H, s), 3.52 (3H, s), 4.09 (2H, m), 4.88 (0.5H, m), 4.93 (0.5H, m), 4.96 (1H, brd, J = 10.3)Hz), 5.05 (1H, m), 5.18 (1H, m), 5.49 (1H, m), 5.59-5.65 (2H, m), 5.99 (1H, d, J = 15.4 Hz), 7.30-7.42 (12H, m), and 7.447.56 (8H, m); ESIMS m/z 1283 (M + Na)⁺; HRESIMS m/z 1283.4760 [calcd for C₆₄H₇₂O₁₂F₁₂Na (M + Na)⁺, 1283.4723].

Tetrakis-(R)-MTPA Ester (3b) of Reduction Product of Amphidinolide W (1). The reduction product of amphidinolide W ($\hat{\mathbf{1}}$, 0.5 mg) was treated with (S)–($\hat{+}$)-MTPACl ($\hat{\mathbf{3}} \mu L$) by the same procedure as described above to afford the tetrakis-(R)-MTPA ester (3b, 0.3 mg) of 1. 3b: colorless oil; ¹H NMR (CDCl₃) δ 0.79 (1.2H, d, J = 6.7 Hz), 0.83 (1.8H, d, J = 6.7 Hz), 0.86 (6H. m), 1.00 (3H, t, J = 7.0 Hz), 1.40-1.65 (4H, m), 1.63 (3H, s), 1.70 (1H, m), 1.81 (1H, m), 1.99 (2H, m), 2.10 (2H, m), 2.21 (0.4H, m), 2.27-2.38 (1.6H, m), 2.44 (1H, m), 3.46 (6H, s), 3.50 (1.2H, s), 3.51 (1.8H, s), 3.52 (3H, s), 3.94 (1H, dd, J = 6.9, 10.7 Hz), 4.12 (1H, dd, J = 5.4 and 10.7 Hz), 4.89 (0.6H, m), 4.96 (1.4H, m), 5.11 (1H, m), 5.17 (0.4H, m), 5.22 (0.6H, m), 5.54 (1H, m), 5.58–5.66 (2H, m), 5.94 (1H, d, J=15.4 Hz), 7.33–7.42 (12H, m), and 7.44-7.57 (8H, m); ESIMS m/z 1283 (M + Na)+; HRESIMS m/z 1283.4764 [calcd for C₆₄H₇₂O₁₂F₁₂Na (M + Na)⁺, 1283.4723]

Baeyer-Villiger Reaction of Amphidinolide W (1). Amphidinolide W (1, 0.5 mg) in MeOH/EtOAc (2:1, 75 μ L) was treated with 5% rhodium-alumina (0.7 mg) under H₂ atmosphere at room temperature for 3 h. After filtration and evaporation of the solvent, the residue was dissolved in CH₂Cl₂ (50 μ L), and treated with TFPA prepared by adding trifluoroacetic anhydride (20 μ L) to a 30% aqueous H₂O₂ (4 μ L) in CH₂-Cl₂ (25 µL) at 4 °C for 12 h. After evaporation, to the reaction mixture were added THF (50 μ L) and LiAlH₄ (2.5 mg), and the mixture was stirred at room temperature for 1 h. The solvent was evaporated, the residue was extracted with EtOAc (200 μ L \times 3), and the extract was concentrated in vacuo to give a crude C-6-C-20 segment (1.1 mg). An aliquot (0.5 mg) of the crude residue was dissolved in CH₂Cl₂ (50 µL), 4-(dimethylamino)pyridine (50 μ g), triethylamine (5 μ L), and (*R*)-(-)-MTPACl (2 μ L) were added to the mixture, and stirring was continued at room temperature for 14 h. After addition of N,N-dimethyl-1,3propanediamine (4 μ L), the solvent was evaporated in vacuo. The residue was passed through a silica gel column (hexane/ EtOAc, 8:1) and then purified by C₁₈ HPLC (TSK-gel ODS-100S, 4.6 × 250 mm; eluent, CH₃CN/H₂O (95:5); flow rate, 1 mL/min; UV detection at 230 nm) to afford a tris-(*S*)-MTPA ester **4a** (0.2 mg, t_R 30 min) of C-6–C-20 segment. **4a**: ¹H NMR (CDCl₃) δ 0.79 (3H, d, J = 6.7 Hz), 0.81 (3H, d, J = 6.7 Hz), 0.84 (3H, t, J = 7.0 Hz), 1.05–1.27 (8H, m), 1.28 (3H, d, J = 6.0 Hz), 1.30–1.84 (12H, m), 3.48 (6H, s), 3.55 (3H, s), 5.07 (1H, m), 5.15 (1H, m), 5.28 (1H, m), 7.33–7.42 (9H, m), and 7.44–7.57 (6H, m); ESIMS m/z 973 (M + Na)⁺; HRESIMS m/z 973.3953 [calcd for C₄₈H₅₉O₉F₉Na (M + Na)⁺, 973.3913].

A tris-(*R*)-MTPA ester **4b** (0.08 mg, $t_{\rm R}$ 27 min) was yielded by treatment of the crude C-6–C-20 segment (0.2 mg) of **1** with (*S*)-(+)-MTPACl. **4b**: ¹H NMR (CDCl₃) δ 0.78 (3H, d, J = 6.7Hz), 0.84 (3H, t, J = 7.0 Hz), 0.85 (3H, d, J = 6.7 Hz), 1.05–.34 (8H, m), 1.36 (3H, d, J = 6.0 Hz), 1.37–1.80 (12H, m), 3.48 (6H, s), 3.50 (3H, s), 5.15 (3H, m), 7.33–7.42 (9H, m), and 7.44–7.57 (6H, m); ESIMS *m*/*z* 973 (M + Na)⁺; HRESIMS *m*/*z* 973.3882 [calcd for C₄₈H₅₉O₉F₉Na (M + Na)⁺, 973.3913].

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Supporting Information Available: Table S1 and spectral data of **1**. This material is available free of charge via the Internet at http://pubs.acs.org.

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